Assignment of the ¹³C nuclear magnetic resonance spectrum of a short DNA-duplex with ¹H-detected two-dimensional heteronuclear correlation spectroscopy

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ABSTRACT
Proton-detected ¹H-¹³C heteronuclear correlated spectroscopy ([¹H, ¹³C]-COSY) was used to establish relations between the carbon-13 and proton nuclear magnetic resonance chemical shifts in the hexadeoxynucleoside pentaphosphate d-(GCATGC)₂. Using the previously established sequence-specific proton NMR assignments, sequence-specific assignments were thus obtained for nearly all proton-bearing carbons. This approach offers a new criterion for distinguishing between the proton NMR lines of purines and pyrimidines, based on the different proton-carbon-13 coupling constants. Furthermore, the adenine ring carbon 2 has a unique carbon-13 chemical shift, which enables ¹₁ and ¹₁ straightforward identification of the adenine C2H resonances by [¹H, ¹³C]-COSY.

INTRODUCTION

Most NMR studies of nucleotides have employed proton and phosphorous resonances, whereas ^{13}C and ^{15}N were little used because of the low sensitivity. With the introduction of heteronuclear 2D NMR experiments with proton detection (1-9), ^{13}C and ^{15}N resonances can now also be assigned conveniently and used to analyze nucleotides without isotope enrichment. In this report we demonstrate the use of heteronuclear correlated spectroscopy ([^{1}H , ^{13}C]-COSY) with studies of the hexadeoxynucleoside pentaphosphate duplex d-(GCATGC)₂:

$$d^{-5}G_1$$
 C_2 A_3 T_4 G_5 C_6 C_6 C_7 C_{-1} G_{-2} C_{-3} C_{-1} C_{-5} C_{-6} C_{-6}

We use an experiment designed for efficient suppression of the signals from carbon-12 bound protons, enabling easy observation of the much weaker signals from the 13 C satellites (1,10,11). Previously, sequence-specific 13 C NMR assignments were obtained with $[^{1}$ H, 13 C]-COSY for some oligoribonucleo-

tides (12) and oligodeoxyribonucleotides (13) in single stranded form. For DNA fragments in duplex form (e.g., d-(CGCGCG)₂, d-(CGCGCGCG)₂; (14)) one-dimensional NMR methods were used and combined with empirical criteria for resonance assignments. The presently described two-dimensional NMR approach has the advantage of superior reliability and speed. It is of particular interest with regard to the potential of 13 C NMR spectroscopy for studies of structure and dynamics of oligonucleotides (15), as well as its use as a complementary method to 1 H NMR spectroscopy for studies of DNA-ligand interactions (e.g.,16).

MATERIALS AND METHODS

The hexadeoxynucleoside pentaphosphate d-GCATGC was synthesized by the phosphotriester method in liquid phase as described elsewhere (17,18). The oligonucleotide was used as the Na $^{+}$ salt (obtained by running the DNA over a Dowex 50W X8 column in the Na $^{+}$ -form, with H $_{2}0$ as the solvent, followed by desalting with Sephadex G-25). The NMR sample was prepared by dissolving the DNA in 500 μ l buffer solution in D $_{2}0$, containing 0.1 M NaCl and 50 mM phosphate, p 2 H = 7.0. This solution was twice lyophylized and redissolved in 500 μ l of 99.96% D $_{2}0$. After a third lyophylization the final solvent used was 500 μ l of 99.996% D $_{2}0$ (Stohler Isotopes). The concentration of the DNA was 20 mM in duplex.

The NMR spectra were recorded in a 5 mm sample tube on a Bruker AM 360 spectrometer at 297 ± 0.5 K. A [1 H, 13 C]-COSY experiment with 490 t₁ values was recorded. The maximum t₁ value was 12.3 ms. Each free induction decay consisted of 1024 data points with a maximum t₂ value of 140 ms. Prior to Fourier transformation the time domain data set was multiplied with a trapezoidal window along t₂ , with the plateau reaching from 3 ms to 82 ms, and a cosine window was applied along ω_1 . Chemical shifts are quoted relative to internal TSP (2,2,3,3-tetradeutero-3-(trimethylsilyl)propionate, sodium salt).

RESULTS AND DISCUSSION

Figure 1 shows a survey of the [1 H, 13 C]-COSY spectrum of the hexadeoxy-nucleoside pentaphosphate d-(GCATGC) in duplex form. The 1 H spectrum is along ω_2 , the 13 C spectrum along ω_1 . For each of the observed cross peaks the two doublet components have opposite sign. The assignments of the different groups of resonance peaks to particular CH or CH $_2$ fragments in the mononucleotides follows from the chemical shift ranges generally observed for the corresponding 13 C and 1 H lines (e.g.,19-22).

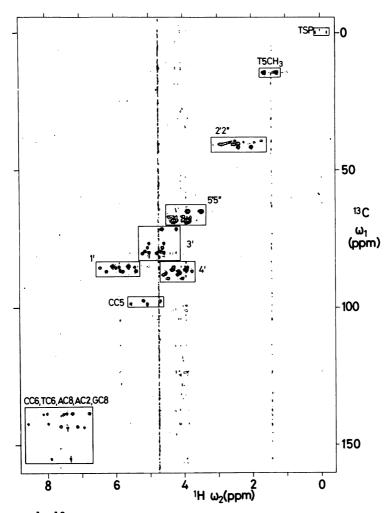
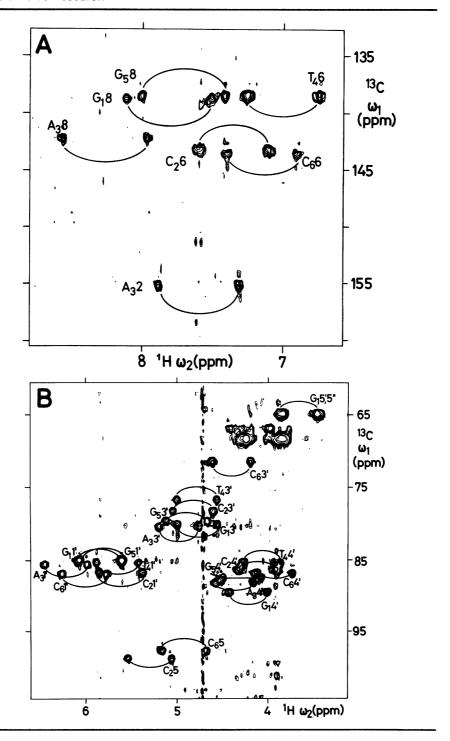


Figure 1: $[^1\text{H},^{13}\text{C}]$ -COSY of a 20 mM solution of d-(GCATGC) in D₂O, containing 0.1 M NaCl, 0.05 M phosphate, pH=7.0, T=297 K. The total measuring time was 72 hours. The different types of cross peaks are boxed and labelled. TSP identifies the cross peak due to the internal reference. The cross peaks show an antiphase doublet splitting along ω_2 due to the heteronuclear coupling. The opposite signs of the two multiplet components (left component positive, right component negative) are not indicated in the figure.

Figure 2 contains expanded plots of three spectral regions in Fig.1. The multiplet splittings along ω_2 , which are due to the $^{13}\text{C-}^{1}\text{H}$ couplings, are indicated with curved lines. Since the ^{1}H NMR spectrum of the hexanucleotide was previously assigned (17), sequence specific assignments for all $^{1}\text{H-bea-}$



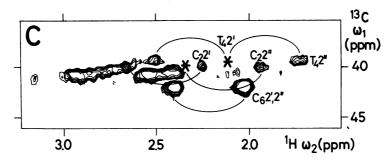


Figure 2: Expanded plots from the spectrum of Fig. 1. (A) Region of the cross peaks corresponding to ring C-H groups of the bases. (B) Region of the cross peaks corresponding to 1'CH, 3'CH, 4'CH, and 5'CH₂ fragments of the deoxyriboses. (C) Region of the cross peaks corresponding to the 2'CH₂ fragments of the deoxyriboses. The heteronuclear splittings of the cross peaks along ω_2 are indicated by curved lines. The peaks are identified by the one-letter abbreviations for the deoxynucleotides and the sequence position. In spectrum (C), stars indicate the positions of multiplet components of cross peaks which were cancelled by overlap (see text).

ring $^{13}\text{C-nuclei}$ were obtained as indicated in the figure, the only exceptions being some C2' and C5' positions. Table 1 lists the chemical shifts and the heteronuclear coupling constants $^1\text{J}_\text{CH}$.

The sequence-specific ¹³C resonance assignments were accomplished by a comparison of the ¹H chemical shifts in the spectrum of Fig.1 with the sequence-specific assignments derived earlier by $^{1}\mathrm{H}$ NOESY and $^{31}\mathrm{P}\text{-relayed}$ $^{1}\mathrm{H}$ COSY (17). This was straightforward for all C-H fragments, but for the CH, groups only partial assignments of the carbon-13 resonances were obtained (Table 1). For all CH₂ groups limitations resulted from cancellation of cross peak components due to mutual overlap, which is especially pronounced in cases where the chemical shift dispersion of the $^{13}\mathrm{C}$ and $^{1}\mathrm{H}$ resonances is small, and for the 5'CH, groups we had further the situation that all 5'H and 5"H resonances had not been assigned (17). In the region of the [1H, 13C]-COSY spectrum containing the cross peaks of the 2'CH₂-groups the cancellation of cross peak multiplet components due to two different reasons can be demonstrated: First, for T_{A} the cancellation of two multiplet components is due to the fact that the positive left hand component of the cross peak between $T_{\Delta}C2'$ and $T_{\Delta}2'H$ overlaps with the negative right hand component of the $T_4C2'-T_42"H$ cross peak. This cancellation thus arises because the difference between the chemical shifts of $T_42'H$ and $T_42''H$ is equal to the coupling constants $^{1}J_{CH}$. Second, for C_{2} the positive left hand component of the higher field cross peak C2'-2"H is cancelled due to overlap

Table 1:	13 _{C-NMR}	chemical	shifts a	and 1	~17	b for	the	hexanucleotide
d-(GCATGC)	2 in 0.1	M NaCl,	0.05 M ph	osphate,	∠n	pH=7.0,	T=29	7 к.

	,					
carbon - 13	G ₁	 c ₂ 	 	 	 G ₅ 	c ₆
			i	i	i	i ————————————————————————————————————
Cl'	85.0	86.8	85.7	85.4	85.0	87.2
C2'	ĺ	40.0	İ	39.5	İ	42.0
C3'	80.2	78.5	80.7	76.9	79.8	71.6
C4'	89.8	86.5	88.3	85.6	87.7	86.8
C5'	65.0			İ	İ	i i
C2			155.4	İ	İ	i i
C5		98.9		İ	İ	97.7
C6		143.4	İ	138.7	İ	144.0
C8	139.4		142.5	İ	138.7	i i
5CH ₃				14.6		i i
j 3				İ	i .	i i
coupling 1 CH						
Cl'-1'H	167	167	165	170	167	172
С3'-3'Н	160	160	160	158	160	158
C4'-4'H	155	150	150	147	150	145
C2 - 2H			204	İ		İ
C5 - 5H		170		1		172
С6 - 6н		178		182		176
C8 - 8H	216		214	İ	210	i i
5CH ₃	ĺ	ĺ		128		l İ
ll				l		

a Chemical shifts are referenced to internal TSP (2,2,3,3-tetradeutero-3-(trimethylsilyl)-propionate, sodium salt) and are within \pm 0.3 ppm.

with the negative right hand component of a 2'CH cross peak of either G_1 , A_3 , or G_5 . The C2' resonances of the latter nucleotides were not individually assigned, because of the very small chemical shift dispersion between 40.5 and 40.9 ppm. For C_6 both protons 2' and 2" have the same chemical shift (17). Only the 5'CH₂ resonances of G_1 were assigned, based on the well-separated, previously assigned proton resonance (17). All other C5' resonances could be located between 67.4 and 69.3 ppm (Fig.2B), but were not further distinguished because of the lack of sequence-specific 1 H NMR assignments (17). No coupling constants for 2'CH₂ or 5'CH₂ are included in Table 1. For the assigned resonances (Fig. 2B, 2C) 1 J_{CH} was estimated to be 143+5 Hz.

A first practical application of the $^{1}\text{H-}^{13}\text{C}$ correlations results from the larger dispersion of the ^{13}C chemical shifts, which enables unambiguous

b 1_{CH} in Hertz + 6 Hertz

 $^{\rm I}$ H NMR assignments in situations where these would be difficult from $^{\rm I}$ H NMR alone. For example, the 13 C resonances of the 4' and 5' carbon atoms are well separated, whereas the corresponding ¹H resonances 4'H, 5'H and 5"H homonuclear ¹H COSY and heteronuclear [¹H, ¹³C]-COSY therefore allows an identification of the protons 4', 5' and 5". Further, the adenine C2 resonances have the lowest field ¹³C shifts of all proton-bearing carbons and are well separated from the other resonances. It thus becomes straightforward to assign the adenine C2-protons, which can otherwise only be identified with relaxation time measurements, deuteration studies, or from NOE's to other protons in the duplex (19,24,25). Haasnoot and coworkers analyzed the $^{13}\mathrm{C}$ NMR spectra of several oligoribonucleotides (12) and proposed a correlation between the ribose ¹³C chemical shifts and parameters connected to conformational equilibria, e.g., the equilibrium, and variations of the backbone torsion angles. For the presently studied hexanucleotide no detailed conformational data are available, so that the validity of these correlations cannot be further checked using this example. It is interesting with regard to future continuation of such studies, however, that the dispersion of the $^{13}\mathrm{C}$ chemical shifts for the sugar carbons 3', 4', and 5' is approximately twice as large as for 1' and 2'.

[1 H, 13 C]-COSY gives also information on the heteronuclear coupling constants 1 J_{CH} (Table 1). The couplings observed here in a double-stranded oligonucleotide are closely similar to those in the corresponding deoxynucleosides (20). The coupling constants 1 J_{CH} are much larger for the purine ring carbons (210±10 Hz) than for the pyrimidine ring carbons (175±8 Hz), so that by [1 H, 13 C]-COSY the ring protons of pyrimidines and purines can be distinguished from the heteronuclear coupling constants 1 J_{CH}. The coupling constants 1 J_{CH} of all 1' positions have about the same values (167±5 Hz), and are quite different from 1 J_{CH} of 3'C (160±5 Hz) or 4'C (150±5 Hz).

Overall, the experiment of Figs.1 and 2 clearly shows that $[^{1}H, ^{13}C]$ -COSY experiments with DNA fragments can be most useful at two different stages of NMR assignments. At the outset of an investigation, the ^{1}H - ^{13}C correlations can help in the identification of the resonances from the different types of hydrogen atoms. Once sequence-specific ^{1}H NMR assignments have been obtained with the now well established procedures (e.g.,19,23-26), sequence-specific ^{13}C NMR assignments can then readily be obtained with these experiments.

In practice the [1H,13c]-COSY experiment can be optimized by adjustment

of the delay time to $\tau = 1/2(^{1}J_{CH})$. Rather than repeating the experiment with a different value of τ for each value of $^{1}J_{CH}$ which occurs in the molecule studied, a compromise for good efficiency may be found by selecting a value corresponding to an average coupling constant. For the present work with DNA the τ value of 3.6 ms corresponds to $^{\rm I}{\rm J}_{\rm CH}$ of 140 Hz. The large deviation of the $^{1}J_{CH}$ values for the ring carbons of purines and pyrimidine from this average value explains the weaker intensity of the corresponding cross peaks when compared to the cross peaks involving ribose CH fragments (Figs. 1 and 2). It is also found that most cross peaks corresponding to CH, fragments have lower intensities than those corresponding to CH, as was also observed in the $[{}^{1}H, {}^{13}C]$ -COSY spectrum of BPTI (11). As was discussed in connection with the cross peaks involving the 2' 13 C resonances (Fig.2C), one reason for this is that the doublets of the cross peaks C2'-2'H and C2'-2"H partially cancel each other, because the negative right hand component of one doublet comes close to, or coincides with the positive left hand component of the other doublet.

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REFERENCES

- Müller, L. (1979) J.Am.Chem.Soc. 101, 4481-4484.
 Maudsley, A.A. and Ernst, R.R. (1977) Chem.Phys.Lett. 50, 368-372.
 Bendall, M.R., Pegg, D.T. and Doddrell, D.M. (1983) J.Magn.Reson. 52, 81-117.
- 4. Bax, A., Griffey, R.H. and Hawkins, B.L. (1983) J.Am.Chem.Soc. 105, 7188-7190.
- 5. Bax, A., Griffey, R.H. and Hawkins, B.L. (1983) J.Magn.Reson. 55, 301-315.
- 6. Frey, M.H., Wagner, G., Vašák, M. Sørensen, O.W., Neuhaus, D., Wörgötter, E., Kägi, J.H.R., Ernst, R.R. and Wüthrich, K. (1985) J.Am.Chem.Soc. 107, 6847-6851.
- 7. Live, D.H., Armitage, I.M., Dalgarno, D.C. and Cowburn, D. (1985) J.Am. Chem. Soc. 107, 1775-1777.
- 8. Live, D.H., Davis, D.G., Agosta, W.C. and Cowburn, D. (1985) J.Am.Chem. Soc. 106, 6104-6105.
- 9. Otyos, J.D., Engeseth, H.R. and Wehrli, S. (1985) J.Magn.Reson. 61,
- 10. Brühwiler, D. and Wagner, G. (1986) J.Magn.Reson., in press.
- 11. Wagner, G. and Brühwiler, D., (1986) Biochemistry, in press.

- 12. Lankhorst, P.P., Erkelens, C., Haasnoot, C.A.G. and Altona, C. (1983) Nucleic Acids Res. 11, 7215-7230.
- 13. Lankhorst, P.P., Haasnoot, C.A.G., Erkelens, C., Westerink, H.P., van der Marel, G.A., van Boom, J.H. and Altona, C. (1985) Nucleic Acids Res. 13, 927-942.
- 14. Borer, P.N., Zanatta, N., Holak, T.A., Levy, G.C., van Boom, J.H. and Wang, A.H.-J. (1984) J.Biomol.Struct.Dyn. 1, 1373-1386.
- 15. Shindo, H. (1981) Eur.J.Biochem. 120, 309-312.
- Berman, E., Brown, S.C., James, T.L. and Shafer, R.H. (1985) Biochemistry 24, 6887-6893.
 Frey, M.H., Leupin, W., Sørensen, O.W., Denny, W.A., Ernst, R.R. and Wüthrich, K. (1985) Biopolymers 24, 2371-2380.
- 18. Denny, W.A., Leupin, W. and Kearns, D.R. (1982) Helv.Chim.Acta 65, 2372-2393.
- 19. Wüthrich, K. (1986) NMR of Proteins and Nucleic Acids, Wiley, New York.
- 20. Chang, C., Da Silva Gomes, J. and Byrn, S.R. (1983) J.Org.Chem. 48, 5151-5160.
- 21. Stothers, J.B. (1972) Carbon-13 NMR Spectroscopy, pp. 468-479, Academic Press, New York.
- 22. Rosenthal, S.N. and Fendler, J.H. (1976) Adv. Phys. Org. Chem. 13, 279-423.
- 23. Feigon, J., Leupin, W., Denny, W.A. and Kearns, D.R. (1983) Biochemistry 22, 5943-5951.
- 24. Scheek, R.M., Russo, N., Boelens, R., Kaptein, R. and van Boom, J.H. (1983) J.Am.Chem.Soc. 105, 2914-2916.
- 25. Hare, D.R., Wemmer, D.E., Chou, S.-H., Drobny, G. and Reid, B.R. (1983) J.Mol.Biol. 171, 319-336.
- 26. Weiss, M.A., Patel, D.J., Sauer, R.T. and Karplus, M. (1984) Proc.Nat. Acad.Sci.USA 81, 130-134.